§ 2.3 Modeling with First Order ODEs Wed. Sept 4 Ourlook: - Recep some steps in formulating a mathematical model with DOFs of - Propose and study some "useful" examples. Physical / Engineering situation: Step 1: Construction of the model \* Common sense \* Known laws \* Experience \* Concentrate on essential features, Step 2: Analysis \* Solutions and its properties, ideally \* Numerical approximation Step 3: Companison with reality + Check (common sense) of Data EXAMPLE 1: Salt tank mixing problems Flor IV \_\_\_\_ V(L): volume of mixture: Vatur 4 sult Q(L): amount of sult in The tant. FLOW OUT  $\bigcirc$ 

lext problem:

"A tank contrains, at time to=0, a quantity Qo lb of salt dissidued in 100 gallons of worker. Water containing 1/4 lb of salt per gallon is entering the tank at a rate of r gallmin, The mixtur is vell-stirred at all times, and drains out of the tank at a rate of 2r gallmin." Questins: • How much mixture is in the bank at a time t, before it empties? Set up an IVP for the amount of salt according to the description. Selve this IVP. 3 Assume r= 3 gal/min and Qo = 0. What is Q(t=10mn)? (1) Volume in the tank:  $\frac{dV}{dt} = FLOW IN - FLOW OUT$   $\frac{dV}{dt} = r - 2r = -r$  $L_{0}V(t) = -rt + C$  and V(0) = 100V(t) = 100 - rt. (2)

Note the tank empties at t= 100/ gter which expressions do not make sense. 2 IVP for sult amount:  $\frac{dQ}{dt} = RATE IN$ RATE OUT  $= \frac{1}{4} \times r - \frac{Q(\ell)}{V(\ell)} \times 2r$   $\int_{0}^{\ell} \int_{0}^{\ell} \int_{0}^{$  $\begin{cases} \frac{dQ}{dt} = \frac{\Gamma}{4} - \frac{2\Gamma}{100 - rt} Q(t) \\ Q(0) = Q_0 \end{cases}$ Solution: integrating factor \* Standard form:  $\frac{dQ}{dt} + \frac{2r}{100-rt}Q(t) = \frac{r}{4}$  p(t) q(t)



\* Integrating factor:  $m(t) = exp\left(\int p(t)\right)$ Here,  $p(t) = \frac{2r}{100-rt}$ To integrate, use  $\int \frac{u'}{u} = \ln(u(\ell))$ Here u(t) = 100 - rt so u'(t) = -rThen,  $\int p(t) = -2 \int \frac{-r}{100 - rt} = -2 \ln(100 - rt)$  $m(t) = \exp\left(-2 \ln(100 - rt)\right) = (100 - rt)^{-2}$ \* Solution:  $Q(t) = \frac{1}{m(t)} \left( \int m(t) q(t) + C \right)$ Now  $m(t)q(t) = \frac{r/4}{(100 - rt)^2}$ To integrate, use  $\int \frac{-u^{\prime}}{u^2} = \frac{1}{u(t)}$ . Here, u(t) = 100 - rt, u'(t) = -r

50 now, 
$$\int p(t)q(t) = \frac{1}{4} \int \frac{-(-r)}{(100 - rt)^2} = \frac{1}{4(100 - rt)^2} = \frac{1}{4(100 - rt)} + C$$
  
 $Q(t) = \frac{100 - rt}{4} + C(100 - rt)^2$   
 $p(t) = \frac{100 - rt}{4} + C(100 - rt)^2$   
 $p(t) = Q_0 = 25 + 100^2 C$   
 $C = 10^{-4} (Q_0 - 25)$   
So...  $Q(t) = \frac{1}{4} (100 - rt) + (\frac{100 - rt}{100})^2 (Q_0 - 25)$   
(3) Numerical value:  $t = 10$  mn,  $r = 3$  gd/mn;  $Q_0 = 0$   
First, chech:  $t = 10$  mn  $< \frac{100}{r} = 33$ ... min.  
 $Q(10 \text{ mn}) = \frac{1}{4} (100 - 30) + (\frac{100 - 30}{100})^2 x(0 - 25)$ 



6 Friday Sept. 7 S2.3 and 2.5 Note: 3 possible cases for self tank problems. . Flow IN > flow OUT Gamonut of mixture in waser until a aflow. Flow OUT > flow IN G amount of mixture devease until empty, , Flaw IN = flow OUT. G amount of mixture 5 constant. Easiest case: DOE has constant coefficients  $\frac{dk}{dt} = c_{in}r_{in} - \frac{r_{out}}{V_o}Q(t)$ See the textbook: example 1, § 2.3 Example 2: Compound interest. Retirement account: regular deposits at fixed rate. \* Balance: S(E), tyeus after start. \* IVP:  $\frac{dS}{dt} = rS + kc$ interest lycen deposit per year (6)

 $\mathcal{E}$ \* Solution: integrating factors. (1)  $\frac{dS}{dt} - rS = k$  (2) m(t) = exp(-rt)(3)  $\frac{a}{dt} \left( e^{-rt}S(t) \right) = k e^{-rt}$  $S(t) = e^{rt} \left( -\frac{h}{r} e^{-rt} + C \right)$ (4) S(t) = Soert + k (ert - 1) initial investment regular deposit + interest accruing or situation. (5) \* On the stock market, "interst rate" is not constant ( predictible. RANDOT. Note \* See texthook for "disordi" model, more precise. (after example 2) Example 3 Temperature models Newton's law of cooling: Rate of change of temperature proportional to difference of temperature her seen the object and its surroundings. (7)

(8) Math model:  $k\left(T_{s}-T(t)\right)$  $\frac{d\tau}{dt} =$ where T(t): temperature of the object,  $T_{s}(t)$ : temperature of its surroundiags, usually constant; k : cooling constant Trate. Selution: If To constant, use separation of nomiables:  $\int \frac{dT}{T-T_c} = -\int k$ Thus  $ln |T-T_s| = -kt + C$  $T-T_s = \frac{\pm e}{c_r} \frac{e^{-ht}}{arbitrary}$  constant Initial temperature  $T(0) = T_0$ :  $T(t) = T_s + (T_o - T_s) e^{-kt}$ Since k > O always (thermodynamics?) we observe  $T(t) \longrightarrow T_s$  (the equilibrium) independently of the original temperature (stability). 18,